



# Pharmaceutical Analytical Chemistry I

الأستاذ الدكتور جمعه الزهوري (دكتوراه صيدلة-ألمانيا 1991 )

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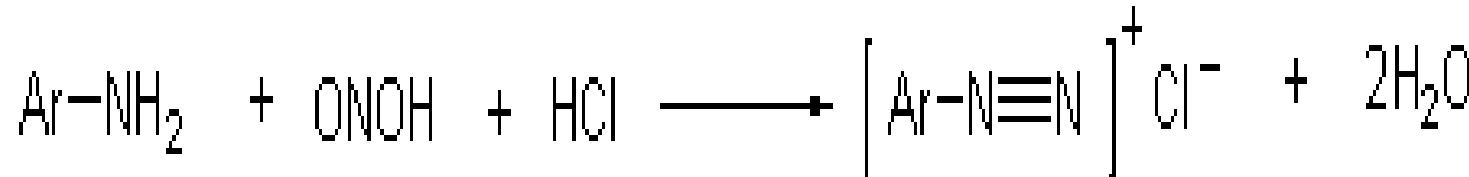
# *Diazometric Titration of drugs*

Prof.Dr.Journal Al-Zehouri

## Diazometric titration of drugs

***Principle :***

**Primary amine drugs react with Nitrous acid in cold to give diazonium salt according to the following reaction :**

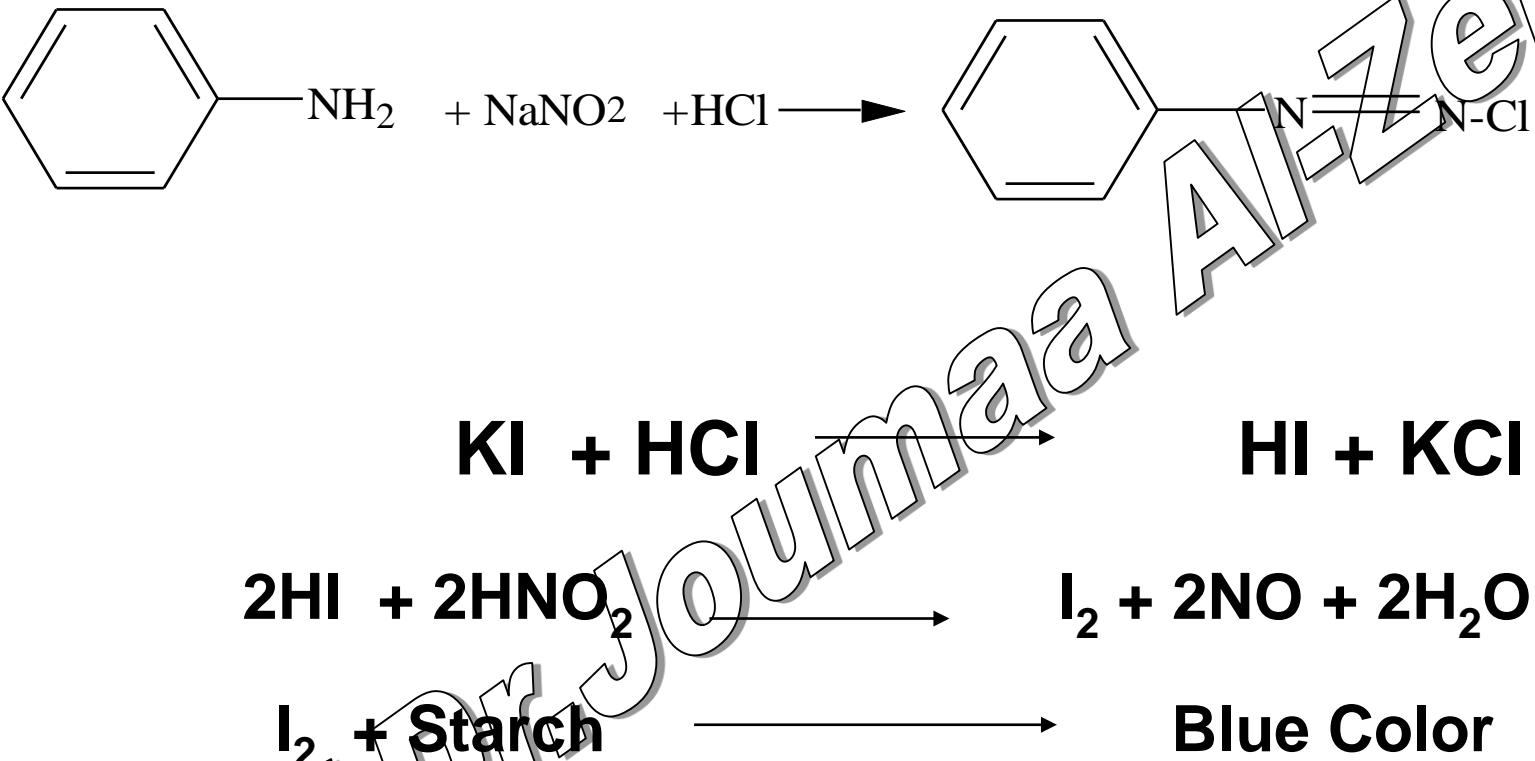


## Diazometric titration of drugs

- *Practical Application :*

In this titration we used Sodium nitrite as primary standard solution after standardization with sulphamic acid ( $\text{H}_3\text{NO}_3\text{S}$ ) and [Starch-KI paste] as indicator to get blue color according to the following reactions.

# Diazometric titration of drugs



تم المعايرة باضافة نتریت الصودیوم حتى تمام التفاعل وأول قطرة زائدة من نتریت الصودیوم (حمض الأزوتی ) يؤکسد الیودور لیود فینتون النشاء

Diazometric  
titration of drugs

# *Pharmaceutical Application I- Raw Material*

Prof.Dr.Jumana  
Prof. Dr. J-Zehouri

## Diazometric titration of drugs

Benzocain	0.1 NaNO <sub>2</sub>	0.01652
Dopsone	0.1 NaNO <sub>2</sub>	0.01242
Procainamide,HCl	0.1 NaNO <sub>2</sub>	0.02718
Procaine,HCl	0.1 NaNO <sub>2</sub>	0.02728
Sulphaacetamide Sodium	0.1 NaNO <sub>2</sub>	0.02362
Sulphadiazine	0.1 NaNO <sub>2</sub>	0.025033
Sulphadimidine	0.1 NaNO <sub>2</sub>	0.02783
Sulphamethoxypyridazine	0.1 NaNO <sub>2</sub>	0.02803
Sulphapyridine	0.1 NaNO <sub>2</sub>	0.02493
Sulphadimidine sodium	0.1 NaNO <sub>2</sub>	0.03003
Sulphamethizole	0.1 NaNO <sub>2</sub>	0.02703
Sulphamethoxine	0.1 NaNO <sub>2</sub>	0.03103
Sulphamethoxypyridazine	0.1 NaNO <sub>2</sub>	0.02803
Sulphapyridine	0.1 NaNO <sub>2</sub>	0.02493

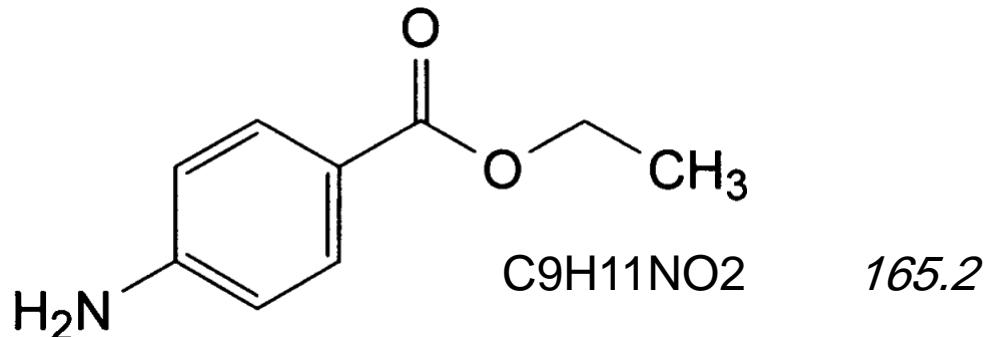
Diazometric  
titration of drugs

# *Pharmaceutical Application II- Dosage Form*

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Chloramphenicol Capsules	0.1 NaNO <sub>2</sub>	0.0327
Co-Trimoxazole (i.v) infusion	0.1 NaNO <sub>2</sub>	0.02533
Dopsone,Tablets	0.1 NaNO <sub>2</sub>	0.01242
Paracetamol Tablets	0.1 NaNO <sub>2</sub>	0.01510
Paediatric Sulphacetamid,Oral	0.1 NaNO <sub>2</sub>	0.02183
Procainamid injection	0.1 NaNO <sub>2</sub>	0.02718
Sulphacetamide, Eye drops	0.1 NaNO <sub>2</sub>	0.02542
Sulphacetamid,Eye Ointiment	0.1 NaNO <sub>2</sub>	0.02542
Sulphadiazine injection	0.1 NaNO <sub>2</sub>	0.02503
Sulphadimidine injection	0.1 NaNO <sub>2</sub>	0.03003
Sulphadimethoxine Tablets	0.1 NaNO <sub>2</sub>	0.03103
Sulphadimidine Tablets	0.1 NaNO <sub>2</sub>	0.02783
Sulphaquanidine ,Tablets	0.1 NaNO <sub>2</sub>	0.0214
Sulphamethizole,Tablets	0.1 NaNO <sub>2</sub>	0.02703
Sulphamethoxy Pyridazine,Tablets	0.1 NaNO <sub>2</sub>	0.02803
Sulphapyridine Tablets	0.1 NaNO <sub>2</sub>	0.02493
Sulphathiazole Tablets	0.1 NaNO <sub>2</sub>	0.02553

## Benzocaine



## Action and use

Local anaesthetic.

## ASSAY

Dissolve 0.400 g in a mixture of 25 ml of *hydrochloric acid R* and 50 ml of *water R*. Carry out the determination of primary aromatic amino-nitrogen (2.5.8).

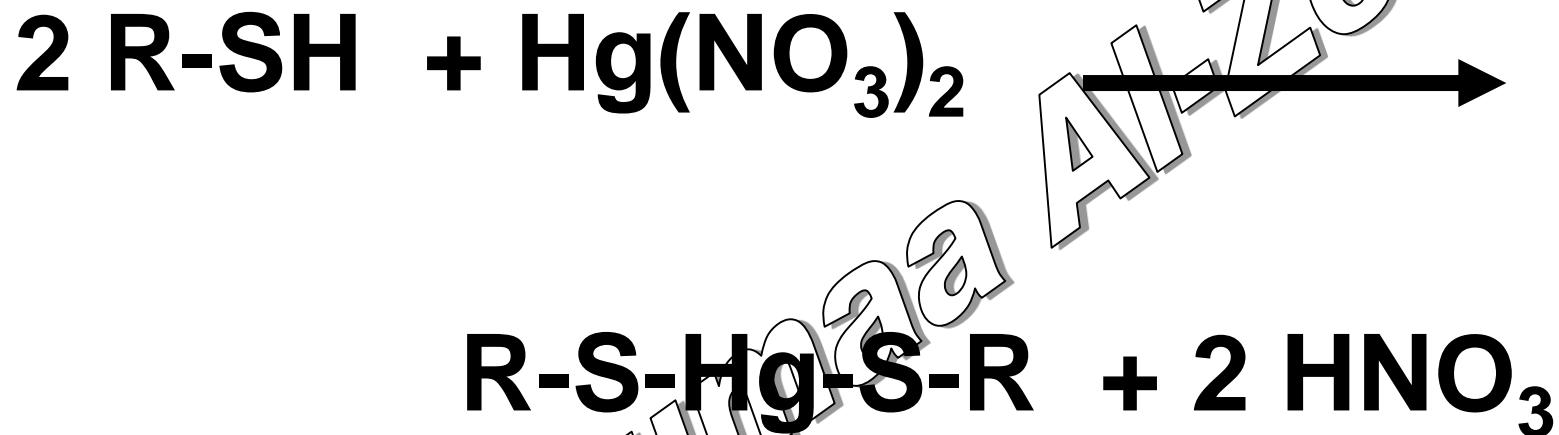
1 ml of 0.1*M* *sodium nitrite* is equivalent to 16.52 mg of  $\text{C}_9\text{H}_{11}\text{NO}_2$ .

# **Mercurimetric Titration of drugs**

**Prof.Dr.-Sohumac Al-Zehouri**

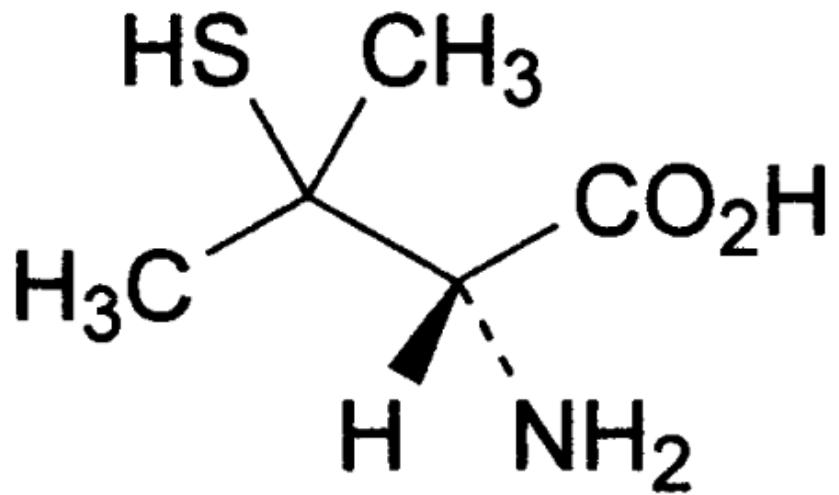
## Principle :

- Mercury (Hg) has the ability to complex two –SH group .
- All drugs which including **Sulphadryl** compounds react quantitative with Mercuric salts.
- Mercury II nitrate used in Pharmacopia as standard solution which reacted with the sulphadryl drugs according to the following reaction :



- Diphenylthiocarbazone (dithizone)  
used as indicator

## Penicillamine



### Action and use

Used in treatment of rheumatoid arthritis and in treatment of lead poisoning.

# Penicillamine Tablets

BP2007

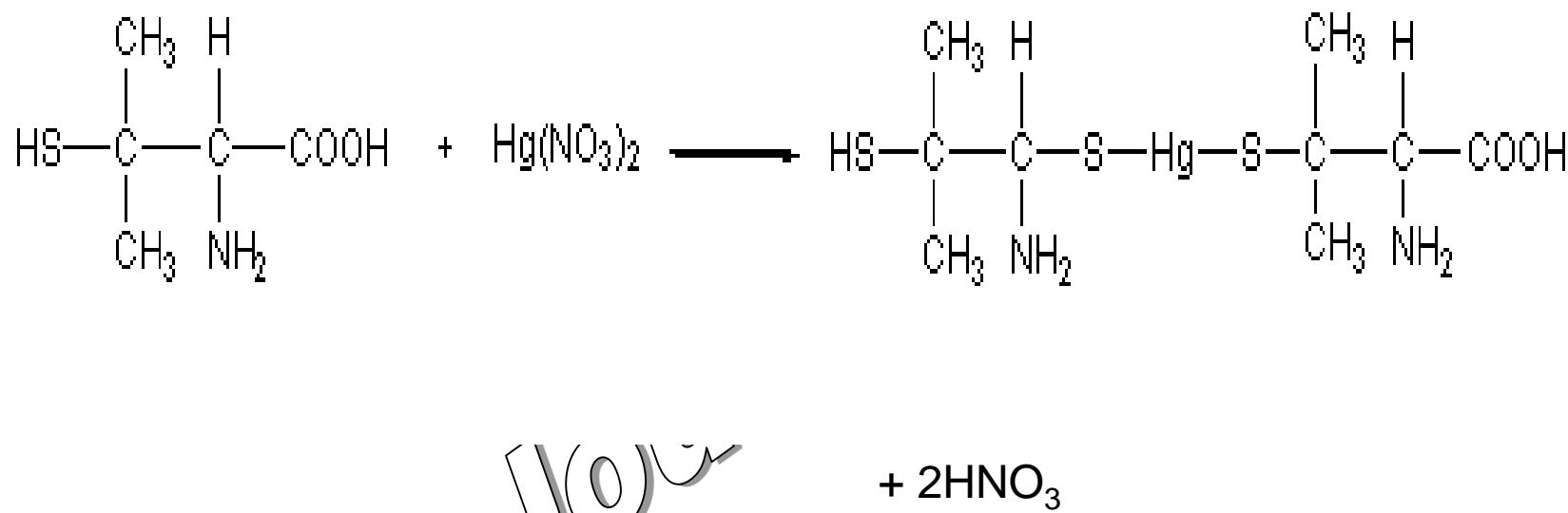
Content of penicillamine, C<sub>5</sub>H<sub>11</sub>NO<sub>2</sub>S

95.0 to 105.0% of the stated amount.

## ASSAY

*Weigh and powder 20 tablets. Dissolve a quantity of the powder containing 0.1 g of Penicillamine as completely as possible in 50 ml of water and filter. Add to the filtrate 5 ml of 1M sodium hydroxide and 0.2 ml of a 0.1% w/v solution of dithizone in ethanol (96%) and titrate with 0.02M mercury(II) nitrate VS. Each ml of 0.02M mercury(II) nitrate VS is equivalent to 5.968 mg of C<sub>5</sub>H<sub>11</sub>NO<sub>2</sub>S.*

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Amoxycilline trihydrate	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.007308
Ampicillin trihydrate	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.006988
Benzathine Penicillin	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.00909
Benzylpenicillin Potassium	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.00745
Benzyl Penicelline	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.007128
Penicillamine	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.00596
Procaine Penicellen	0.02 M Hg(NO <sub>3</sub> ) <sub>2</sub>	0.01141

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Penicillamine, Tablets & Capsules	0.02 M $\text{Hg}(\text{NO}_3)_2$	0.05968
Propylthiouracil Tablets	0.02 M $\text{Hg}(\text{NO}_3)_2$	0.006808

Prof.Dr.JOURNAL



**Thank you**

Prof.Dr.Joumaa Al-Zehouri

**Q & A**